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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.005 Å R factor = 0.047 wR factor = 0.122 Data-to-parameter ratio = 21.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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The crystal structure of the title molecule, $C_{13}H_{17}BrO_2$, contains two crystallographically independent molecules in the asymmetric unit. The cyclohexane rings of these two molecules adopt chair conformations. In the solid state, the

molecules are aggregated around the cell corners to form a

cooperative

trans-2-(2-Bromo-4-methylphenyloxy)cyclohexanol

 $O-H\cdots O-H\cdots O-H\cdots$

Comment

four-membered

hydrogen-bonded ring.

The cyclohexanol and its derivatives were proven to be an important tool in both biochemical and physiological studies of the cholinergin nerve terminal (Rogers *et al.*, 1989). Also, many of the cyclohexanol derivatives exhibit good receptor properties against the inhibition of acetylcholine storage by nerve terminal synaptic vesicles (Marshall & Parsons, 1987). The crystal structure determination of the title compound, (I), one of the above derivatives, was performed in order to elucidate its molecular conformation.



The asymmetric unit of (I) contains two crystallographically independent molecules linked by an O2B-H2B···O2A hydrogen bond (Table 1). No significant differences in the corresponding bond lengths and angles of these two molecules are observed and they show normal values. The cyclohexane ring in both molecules adopts the chair conformation and the hydroxyl and benzoyl groups are equatorially attached. In the solid state, the inversion-related molecules (symmetry code: 2-x, -y, 2-z) are linked by $O-H \cdots O$ hydrogen bonds to form molecular aggregates with a four-membered cooperative $O-H \cdots O-H \cdots O-H \cdots$ hydrogen-bonded ring. The crystal structure is further stabilized by weak $C-H \cdots \pi$ interactions involving the phenyl rings of molecule A (C_gA = centroid of C1A-C6A) and molecule B (C_gB = centroid of C1B-C6B).

Experimental

To a mixture of cyclohexene oxide (0.5 g, 5 mmol) and neutral alumina (3.5 g) in dry benzene (50 ml) was added dropwise 2-bromo-4-methylphenol (0.95 g, 5 mmol) at room temperature. It was further refluxed until disappearance of the starting material in TLC. Then it was filtered and the solvent was removed under vaccum. The residue

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Figure 1

The structure of the asymmetric unit of (I) showing 50% probability displacement ellipsoids and the atom-numbering scheme.

obtained was further purified by column chromatography using silica gel to afford the title compound in good yield (1.25 g, 86%). Single crystals were grown by slow evaporation of the solvent from a solution of the compound in chloroform–methanol.

Z = 4

 $D_x = 1.462 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 4186

reflections

 $\mu = 3.16 \text{ mm}^{-1}$

T = 293 (2) K Plate, colourless $0.24 \times 0.20 \times 0.10$ mm

 $R_{\rm int} = 0.030$

 $\theta_{\rm max} = 28.3^\circ$

 $\begin{array}{l} h=-9 \rightarrow 12 \\ k=-14 \rightarrow 15 \end{array}$

 $l = -16 \rightarrow 15$

6185 independent reflections

3937 reflections with $I > 2\sigma(I)$

 $\theta = 1.7 {-} 28.3^{\circ}$

Crystal data

$C_{13}H_{17}BrO_2$
$M_r = 285.18$
Triclinic, P1
a = 9.7151 (2) Å
b = 11.6911 (3) Å
c = 12.4332(3) Å
$\alpha = 77.401 \ (1)^{\circ}$
$\beta = 80.521 \ (1)^{\circ}$
$\gamma = 70.905 \ (1)^{\circ}$
$V = 1295.65 (5) \text{ Å}^3$
Data collection

Siemens SMART CCD areadetector diffractometer ω scans Absorption correction: empirical (*SADABS*; Sheldrick, 1996) *T*_{min} = 0.518, *T*_{max} = 0.743 9000 measured reflections

Refinement

Refinement on F^2	H-atom parameters constrained		
$R[F^2 > 2\sigma(F^2)] = 0.047$	$w = 1/[\sigma^2(F_o^2) + (0.0513P)^2]$		
$wR(F^2) = 0.122$	where $P = (F_o^2 + 2F_c^2)/3$		
S = 1.00	$(\Delta/\sigma)_{\rm max} = 0.001$		
6185 reflections	$\Delta \rho_{\rm max} = 0.42 \text{ e} \text{ Å}^{-3}$		
293 parameters	$\Delta \rho_{\rm min} = -0.63 \text{ e} \text{ Å}^{-3}$		

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O2B-H2B\cdots O2A$	0.82	1.97	2.712 (4)	151
$O2A - H2A \cdots O2B^{i}$	0.82	1.98	2.777 (4)	164
$C4A - H4A \cdots C_{\rho}B$	0.93	3.21	3.968 (4)	140
$C12A - H12A \cdot \cdot \cdot C_g A^{ii}$	0.97	3.32	4.272 (5)	166

Symmetry codes: (i) 2 - x, -y, 2 - z; (ii) 2 - x, 1 - y, 1 - z.



Figure 2

A plot of the molecular aggregates of (I), viewed down the b axis.

After checking their presence in a difference map, all the H atoms were placed at geometrically calculated positions and a riding model was used for their refinement; rotating group refinement was used for the methyl and hydroxyl groups.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 1990).

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References

Bruker (1998). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

- Marshall, I. G. & Parsons, S. M. (1987). Trends Neurosci. 10, 174.
- Rogers, G. A., Parson, S. M., Anderson, D. C., Nilsson, L. M., Batir, B. A., Kornreich, W. D., Kaufman, R., Jacobs, R. S. & Kirtman, B. (1989). *J. Med. Chem.* 32, 1217–1230.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Spek, A. L. (1990). Acta Cryst. A46, C-34.